

## Direct observation of sub-domains in the GO single layer

Narae Son<sup>1,5</sup>, Hyunsoo Lee<sup>2</sup>, Tae Gun Kim<sup>3</sup>, Hu Young Jeong<sup>4</sup>, Jong Yun Kim<sup>1</sup>, Gyeongsook Bang<sup>1</sup>,  
Sehun Kim<sup>5</sup>, Jeong Young Park<sup>2</sup> and Sung-Yool Choi<sup>1</sup>

Department of Electrical Engineering and Graphene Research Center, KAIST, 291 Daehak-ro, Daejeon,  
Korea

Graduate School of EEWS and KI for the Nanocentury, KAIST, 291 Daehak-ro, Daejeon, Korea  
Korea Research Institute of Standards and Science, 267 Gajeong-ro, Daejeon 305-340, South Korea  
School of Mechanical & Advanced Materials Engineering and UNIST Central Research Facilities,  
UNIST, 100 Banyeon-ri, Eonyang-eup, Ulsan, Korea

Department of Chemistry and Molecular-Level Interface Research Center, KAIST, 291 Daehak-ro,  
Daejeon, Korea

sungyool.choi@kaist.ac.kr and jeongypark@kaist.ac.kr

### Abstract

Chemically synthesized graphene oxide (GO) is an oxidized single graphitic monolayer that has traditionally served as a precursor for graphene, but has a considerable potential of application for its own characteristics. It has been experimentally proved that GO monolayer consists of randomly distributed regions, aromatic regions ( $sp^2$  carbon atoms) and oxygenated aliphatic regions ( $sp^3$  carbon atoms) by raman spectroscopy [1], scanning tunneling microscopy (STM) [2] and high-resolution transmission electron microscopy (HR-TEM) [3]. However, these methods have disadvantages such as complicated and time-consuming preparation process and narrow region (a few nanometer) being measured.

Here, we investigated the structural, chemical, and electrical properties of synthesized GO simple by using a conductive atomic force measurement (C-AFM) analysis in a few micrometer range. GO monolayer is composed of two different sub-domains, the first one corresponding to a high friction and low conductance domain ( $sp^3$  carbon domain) and the other corresponding to a low friction and high conductance domain ( $sp^2$  carbon domain) [4-5]. At each position of two domains, local I-V characteristics also showed different behaviors depending on the chemical properties. To confirm presence of sub-domains of GO, scanning transmission electron microscopy (STEM), HR-TEM analysis, and oxygen electron energy loss spectroscopy (EELS) mapping were conducted.

### References

- [1] C. Mattevi et al., *Adv. Funct. Mater.*, **19** (2009) p. 2577.
- [2] M. Ishigami, J. H. Chen, W. G. Cullen, M. S. Fuhrer and E. D. Williams, *Nano Lett.*, **7** (2007) p. 1643.
- [3] N. R. Wilson et al., *ACS Nano*, **3** (2009), p. 2547.
- [4] J. H. Ko, S. Kwon, I. S. Byun, J. S. Choi, B. H. Park, Y. H. Kim and J. Y. Park, *Tribol Lett.*, **50** (2013) p. 137.
- [5] S. Seo, C. Jin, Y. R. Jang, J. Lee, S. Kyu. Kim and H. Lee, *J. Mater. Chem.*, **21** (2011) p. 5805.

Figures

